77. H_2SO_3
78. $Ba(OH)_2$
79. PbS
80. NaH_2PO_4
81. $NH_4C_2H_3O_2$
82. Ag_3N
83. SiI_4
84. $ZnCO_3$
85. H_3PO_3
86. SnI_4
87. $Pb(NO_3)_2$
88. NaF
89. $KAl(SO_4)_2$
90. KUO_4
91. $SmCl_3$
92. K_2S_5
93. $Fe_3[Fe(CN)_6]_2$
94. $PtCl_2$
95. PtI_4
96. NI_3
97. $MoCl_5$
98. $La(NO_3)_3$
99. Dy_2O_3
100. V_2O_5

II. Write the correct formula for each of the following compounds:

1. sulfuric acid
2. sodium hydroxide
3. sodium bromide
4. barium hydroxide
5. calcium oxide
6. hydrosulfuric acid
7. lithium sulfate
8. carbon monoxide

(continued)

9. manganese dioxide
10. sulfur dioxide
11. iron (II) sulfate
12. hypochlorous acid
13. potassium permanganate
14. silver chloride
15. copper (II) hydroxide
16. ammonium sulfide
17. nickel bromide
18. iron (II) oxide
19. bromic acid
20. ammonium bisulfate
21. mercury (I) sulfate
22. iron (III) oxide
23. magnesium phosphate
24. nickel bicarbonate
25. zinc hydroxide
26. hydriodic acid
27. diphosphorus pentoxide
28. aluminum phosphate
29. hydrogen acetate
30. copper (II) nitrite
31. nitrogen dioxide
32. phosphorus trichloride
33. sodium phosphate
34. potassium carbonate
35. phosphoric acid
36. lead (IV) chloride
37. tin (II) bromide
38. ammonium hydroxide
39. periodic acid
40. iron (II) hydroxide
41. carbon dioxide
42. dinitrogen pentoxide
43. silver oxide
44. aluminum nitride
45. manganese (II) hydroxide
46. ammonium carbonate
47. aluminum oxide
48. antimony pentasulfide

(continued)